

Cations

Anions

	Ammonium	NH ₄ ⁺		Acetate	C ₂ H ₃ O ₂ ⁻
	Cesium	Cs ⁺		Bromide	Br ⁻
	Copper (I) (Cuprous)	Cu ⁺		Chloride	Cl ⁻
	Hydrogen	H ⁺		Cyanide	CN ⁻
1+	Lithium	Li ⁺		Fluoride	F ⁻
	Potassium	K ⁺		Hydride	H ⁻
	Silver	Ag ⁺		Hydrogen carbonate (bicarbonate)	HCO ₃ ⁻
	Sodium	Na ⁺		Hydrogen sulfate (bisulfate)	HSO ₄ ⁻
			1-	Hydrogen sulfite (bisulfite)	HSO ₃ ⁻
2+	Barium	Ba ²⁺		Hydroxide	OH ⁻
	Cadmium	Cd ²⁺		Iodide	I ⁻
	Calcium	Ca ²⁺		Nitrate	NO ₃ ⁻
	Cobalt (II) (Cobaltous)	Co ²⁺		Permanganate	MnO ₄ ⁻
	Copper (II) (Cupric)	Cu ²⁺		Hypochlorite	ClO ⁻
	Iron (II) (Ferrous)	Fe ²⁺		Chlorite	ClO ₂ ⁻
	Lead (II) (Plumbous)	Pb ²⁺		Chlorate	ClO ₃ ⁻
	Magnesium	Mg ²⁺		Perchlorate	ClO ₄ ⁻
	Manganese (II)	Mn ²⁺			
	Mercury (I) (Mercurous)	Hg ₂ ²⁺		Carbonate	CO ₃ ²⁻
	Mercury (II) (Mercuric)	Hg ²⁺		Chromate	CrO ₄ ²⁻
	Nickel (II) (Nickelous)	Ni ²⁺		Dichromate	Cr ₂ O ₇ ²⁻
	Strontium	Sr ²⁺		Oxalate	C ₂ O ₄ ²⁻
	Tin (II) (Stannous)	Sn ²⁺	2-	Oxide	O ²⁻
	Zinc	Zn ²⁺		Peroxide	O ₂ ²⁻
				Sulfate	SO ₄ ²⁻
				Sulfide	S ²⁻
				Sulfite	SO ₃ ²⁻
3+	Aluminum	Al ³⁺			
	Chromium (III) (Chromic)	Cr ³⁺			
	Iron (III) (Ferric)	Fe ³⁺	3-	Phosphate	PO ₄ ³⁻